# Total synthesis of (+)-nostocyclamide ${ }^{1}$ 

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#### Abstract

The synthesis of (+)-nostocyclamide from the oxazole 1 and thiazoles 2 and 3 is described. The oxazole amino ester 1 b was prepared from $N$-protected alaninamide using a rhodium(II) catalysed $\mathrm{N}-\mathrm{H}$ insertion reaction as a key step, and the thiazoles 2 and 3 were obtained using a modified Hantzsch reaction. The synthesis was completed in six further steps in which fragments 1 and 2 were coupled using mixed anhydride methodology to give the oxazole-thiazole 13, deprotection of which and coupling to 3 gave the linear bis-thiazole oxazole 15. Macrocyclisation using the pentafluorophenyl ester method gave $(+)$-nostocyclamide; the synthesis confirms that the natural product is the $(+)$-enantiomer and has the ( $2 S, 12 R$ ) absolute configuration.


## Introduction

The nitrogen fixing cyanobacteria of the genus Nostoc are a rich source of biologically active substances of diverse structural types. ${ }^{2-11}$ In a detailed survey based on antibacterial activity, the major bioactive component of one strain, Nostoc sp. 31, was shown to be nostocyclamide, a novel macrocyclic peptide consisting of an 18-membered ring containing two thiazoles and one oxazole. ${ }^{12}$ The structure determination was based on extensive ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR analysis and was confirmed by X-ray crystallography, and, although the absolute configuration could not be assigned with complete certainty, the $(2 S, 12 R)$ enantiomer was deemed most likely. In view of our interest in the synthesis of heterocyclic natural products containing oxazoles ${ }^{13,14}$ or thiazoles, ${ }^{15}$ we now report the details of the first synthesis of $(2 S, 12 R)$-nostocyclamide. ${ }^{1}$


## Results and discussion

Simple disconnection of the three amide bonds in nostocyclamide as indicated in Scheme 1 revealed an oxazole amino ester $\mathbf{1}$ and two thiazole derived amino acid derivatives $\mathbf{2}$ and $\mathbf{3}$ as suitable precursors to the natural product. Since the absolute configuration of nostocyclamide had not been assigned with complete certainty, ${ }^{12}$ we elected to synthesise the 'most likely' $(2 S, 12 R)$-enantiomer, thereby dictating that the heterocyclic amino acids $\mathbf{1}$ and $\mathbf{2}$ should be derived from ( $S$ )-alanine and $(R)$-valine respectively. The projected order of coupling of the fragments (Scheme 1) in part dictated the choice of carboxy and amino protecting groups, although two $N$-protecting groups ( Cbz and Boc ) were considered for the valine-derived thiazole 2.

The 5-methyloxazole-4-carboxylic acid $\mathbf{1}$ was readily pre-

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pared using our rhodium catalysed $\mathrm{N}-\mathrm{H}$ insertion reactions of diazo keto esters as described in the preceding paper. ${ }^{16}$ Rhodium(II) acetate catalysed reaction of N -Cbz-protected alaninamide 4 with methyl diazoacetoacetate resulted in chemoselective $\mathrm{N}-\mathrm{H}$ insertion to give the keto amide 5, cyclodehydration of which using the triphenylphosphine-iodine-triethylamine protocol ${ }^{17}$ gave the $N$-protected oxazole ester 6. Finally deprotection by catalytic hydrogenolysis over palladium-on-carbon gave the desired oxazole amino ester $\mathbf{1}$ (Scheme 2).

The synthesis of optically active thiazoles has been widely investigated of late. ${ }^{18-21}$ In particular, Meyers has reported a modification of the Hantzsch reaction, ${ }^{22}$ originally adapted by Holzapfel, ${ }^{23}$ as a reliable route to enantiomerically pure thiazole amino acid derivatives, and successfully applied it in his synthesis of bistratamide $\mathrm{C}^{24}$ a macrocyclic bis-thiazole-oxazole structurally similar to nostocyclamide. Initially the synthesis of the Cbz-protected thiazole 2a was investigated. Hence the thioamide $\mathbf{8 a}$, prepared in modest yield from $N$-Cbz-protected $(R)$-valinamide 7a, was reacted with ethyl bromopyruvate at $-15{ }^{\circ} \mathrm{C}$, followed by dehydration of the resulting hydroxythiazoline with trifluoroacetic anhydride (TFAA) and 2,6lutidine $(2,6-\mathrm{Lu})$ to give the thiazole-4-ester $9 \mathbf{a}$ in good yield





5
${ }_{\mathrm{Et}_{3} \mathrm{~N}}^{\mathrm{P}_{3} \mathrm{P}}$
$\mathrm{CH}_{2} \mathrm{Cl}_{2}$
${ }_{(66 \%)}^{\mathrm{CH}_{2} \mathrm{Cl}_{2}}$


6

Scheme 2
and in $97-98 \%$ ee as determined by HPLC analysis on a chiral column [Chiracel OD column, eluting with hexane-propan-2ol $(9: 1)$ at $\left.2.0 \mathrm{ml} \mathrm{min}{ }^{-1}\right]$. Hydrolysis with sodium hydroxide in aqueous THF gave the required thiazole amino acid 2a (Scheme 3). However the poor yield in the formation of the



$\mathrm{O}_{3}$, DME, $-15^{\circ} \mathrm{C}$ $\mathrm{R}=\mathrm{Cbz}(100 \%)$
$\mathrm{R}=\mathrm{Boc}(73 \%)$


Scheme 3 ( $\mathbf{a}, \mathrm{R}=\mathrm{Cbz} ; \mathbf{b}, \mathrm{R}=\mathrm{Boc}$ )
thioamide 8a, together with difficulties encountered in scaling up the reactions and in subsequent removal of the Cbz-group (see below), led us to investigate the alternative thiazole $\mathbf{2 b}$ in which the side chain nitrogen was protected by a Boc group. The synthesis proceeded as before (Scheme 3), and in general the compounds were easier to purify than the corresponding $N-\mathrm{Cbz}$ derivatives. The thiazole ester $\mathbf{9 b}$ was formed essentially as a single enantiomer [ $>99 \%$ ee by HPLC; Chiracel OD column, eluting with hexane-propan-2-ol $(98: 2)$ at 0.75 ml $\left.\min ^{-1}\right] .{ }^{25}$ Finally alkaline hydrolysis gave the desired thiazole-4carboxylic acid $\mathbf{2 6}{ }^{26}$

Although racemisation was not an issue in the preparation of the glycine derived thiazole amino acid $\mathbf{3}$, a compound previously used in the synthesis of dolastatin, ${ }^{27}$ the compound was prepared from $N$-Boc-glycinamide 10 via the corresponding thioamide $\mathbf{1 1}$ and thiazole ester $\mathbf{1 2},{ }^{28}$ in exactly the same way as the valine analogue (Scheme 4).




1. $\mathrm{KHCO}_{3}, \mathrm{DME},-15^{\circ} \mathrm{C}$ 2. TFAA, $2,6-\mathrm{Lu},-10^{\circ} \mathrm{C}$ (61\%)


## Scheme 4

The coupling of the heterocyclic amino acid residues to form the linear 'peptide' for macrocyclisation was carried out using the mixed anhydride method ( $c f$. ref. 24). Thus reaction of the acid 2a or $\mathbf{2 b}$ with isobutyl chloroformate in the presence of N -methylmorpholine (NMM), followed by addition of the amine 1 gave the $N$-Cbz- or $N$-Boc-protected thiazole-oxazole 13a and 13b in 69 and $71 \%$ yield respectively (Scheme 5). At this stage the N -Cbz-aminoalkyl thiazole series was finally abandoned when we were unable to deprotect the thiazoleoxazole 13a by catalytic hydrogenolysis over palladium-oncarbon or over Pearlman's catalyst. The $N$-Boc derivative 13b, however, was cleanly deprotected using methanolic hydrogen chloride, and the resulting amine $\mathbf{1 4}$ was coupled to the thiazole acid $\mathbf{3}$ using the mixed anhydride method to give the terminally protected linear oxazole-bis-thiazole 15 (Scheme 5).

Although there are several methods available for macrolactamisation, the ring closure protocol used by Schmidt and co-workers has consistently given satisfactory results in a number of examples. ${ }^{29-33}$ Hence the ester 15 was hydrolysed to the corresponding acid 16, and converted into the corresponding pentafluorophenyl ester 17 by coupling with pentafluorophenol in the presence of 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDCI). The pentafluorophenyl ester $\mathbf{1 7}$ was not purified, but underwent deprotection at the N -terminus on treatment with HCl in dioxane, and this was followed by macrolactamisation in a two-phase chloroformaqueous base system to give the macrocyclic peptide nostocyclamide in $74 \%$ yield after chromatography. The synthetic material had mp $259-260^{\circ} \mathrm{C}$ (decomp.) [lit., ${ }^{12} 255.8-256.9^{\circ} \mathrm{C}$ (decomp.)], and a specific rotation of $[\alpha]_{\mathrm{D}}^{19}+51.3$ (c 0.84 , $\mathrm{CHCl}_{3}$ ), together with ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra consistent with those reported for the natural product, although the published ${ }^{13} \mathrm{C}$ NMR data do not list the resonance for $\mathrm{C}-8$, which was observed at $\delta 165.6\left(\mathrm{CDCl}_{3}\right)$ in our synthetic material. Subsequent correspondence with the authors of the original paper confirmed that the natural product does exhibit an additional quaternary carbon signal in its ${ }^{13} \mathrm{C}$ NMR spectrum at $\delta 166.2$ $\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ DMSO) ${ }^{34}$ Although no rotation was reported for nostocyclamide in the original paper due to lack of material, ${ }^{12}$ reisolation of a small amount of the natural product resulted in material with a $(+)$-rotation of $[a]_{\mathrm{D}}^{20}+25\left(\mathrm{CHCl}_{3}\right){ }^{35}$ In view of the discrepancy in the optical rotations, that of the synthetic material was checked on three different samples on two different polarimeters and found to be consistently +53 $( \pm 2)$. The lower value for the natural product may be a result

 13b (R $=\mathrm{Boc})$
$\mathrm{AcCl}, \mathrm{MeOH}$
$(100 \%)$


$\mathrm{C}_{6} \mathrm{~F}_{5} \mathrm{OH}, \mathrm{EDCI}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$


Scheme 5
of the small sample used. Despite the difference in optical rotations, we believe that our synthesis demonstrates that natural $(+)$-nostocyclamide does indeed have the $(2 S, 12 R)$.

## Experimental

For general experimental points, see the preceding paper.
( $S$ )-Methyl 2-(1-aminoethyl)-5-methyloxazole-4-carboxylate 1 A suspension of palladium-on-charcoal $(10 \% ; 150 \mathrm{mg})$ in a solution of ( $S$ )-methyl 2-[(1-benzyloxycarbonylamino)ethyl]-5-methyloxazole-4-carboxylate $\mathbf{6}^{16,36}(0.578 \mathrm{~g}, 1.82 \mathrm{mmol})$ in methanol $(40 \mathrm{ml})$ was stirred under $\mathrm{H}_{2}(1 \mathrm{~atm})$ at room temperature for 4 h . The mixture was filtered through a plug of Celite, the plug was washed with ethyl acetate $(40 \mathrm{ml})$ and the filtrates were combined and evaporated in vacuo to afford the title compound $\mathbf{1}(0.32 \mathrm{~g}, 96 \%)$ as a colourless oil (Found: $\mathrm{M}^{+}, 184.0848$ $\mathrm{C}_{8} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $M, 184.0848$ ); $[a]_{\mathrm{D}}^{23.5}-17.4\left(c 1.15, \mathrm{CHCl}_{3}\right)$; $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3382,3320,1721,1624,1581,1443,1352$, 1197, 1183, 1101 and 1056; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 4.14(1 \mathrm{H}, \mathrm{q}$, $J 6.8, \mathrm{NCHMe}), 3.89(3 \mathrm{H}, \mathrm{s}, \mathrm{MeO}), 2.60(3 \mathrm{H}, \mathrm{s}, 5-\mathrm{Me}), 1.80$ $\left(2 \mathrm{H}, \mathrm{br}\right.$ s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{NH}_{2}\right)$ and $1.49(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{MeCH})$;
$\delta_{\mathrm{c}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 166.55(\mathrm{C}), 163.12(\mathrm{C}), 156.77(\mathrm{C})$, $127.42(\mathrm{C}), 52.26(\mathrm{Me}), 45.91(\mathrm{CH}), 21.94(\mathrm{Me})$ and $12.30(\mathrm{Me})$; $\mathrm{m} / \mathrm{z}$ (EI) $185\left(\mathrm{MH}^{+}, 27 \%\right), 184\left(\mathrm{M}^{+}, 18\right), 169(45), 168(18), 137$ (30), 110 (20), 58 (20) and 44 (100).

## Preparation of thiazole-4-carboxylic acids

(R)-2-[1-(Benzyloxycarbonylamino)-2-methylpropyl]thiazole-4carboxylic acid 2a
(R)- $\boldsymbol{N}^{2}$-(Benzyloxycarbonyl)valinamide 7a. Prepared in $69 \%$ yield by the method described for the ( $S$ )-enantiomer in the preceding paper, ${ }^{16} \mathrm{mp} 210-211{ }^{\circ} \mathrm{C}$ (methanol-water) $\left[\right.$ lit., ${ }^{37}(S)$ enantiomer, $\left.\mathrm{mp} 212^{\circ} \mathrm{C}\right] ;[a]_{\mathrm{D}}^{19}-25.6^{\circ}$ (c 0.74, DMF) $\left\{\right.$ lit., ${ }^{37}$ ( $S$ )-enantiomer [ $\left.a]_{\mathrm{D}}^{25} 22.6^{\circ}(c 1, \mathrm{DMF})\right\}$.
( $R$ )- $N^{2}$-(Benzyloxycarbonyl)thiovalinamide 8a. A solution of the above amide $7 \mathbf{a}(0.568 \mathrm{~g}, 2.27 \mathrm{mmol})$ and Lawesson's reagent $(0.505 \mathrm{~g}, 1.25 \mathrm{mmol})$ in dry dichloromethane ( 50 ml ) was stirred for 16 h at room temperature. The mixture was evaporated in vacuo and purified by flash chromatography on silica, eluting with ether-light petroleum ( $2: 1$ ), to afford the title compound 8a ( $0.25 \mathrm{~g}, 42 \%$ ) as a colourless foam (Found: $\mathrm{M}^{+}, 266.1098 . \mathrm{C}_{13} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}$ requires $\left.M, 266.1089\right) ;[a]_{\mathrm{D}}^{19}+39.2^{\circ}$ (c 2.84, $\mathrm{CHCl}_{3}$ ); $v_{\max }($ film $) / \mathrm{cm}^{-1} 3303,3195,3068,3034,2965$, 2936, 2874, 1699, 1633, 1513, 1454, 1436, 1260, 1229, 1113, 1028,736 and $696 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.21\left(1 \mathrm{H}, \mathrm{s}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$, CSNHH), 7.87 ( 1 H , s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CSNH} H\right), 7.33(5 \mathrm{H}, \mathrm{m}$, $\mathrm{PhH}), 5.76\left(1 \mathrm{H}, \mathrm{d}, J 8.1\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OCONH}\right), 5.10(1 \mathrm{H}, \mathrm{d}$, $J$ 12.3, PhCHH), $5.05(1 \mathrm{H}, \mathrm{d}, J 12.3$, $\mathrm{PhCH} H), 4.29(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C} H \mathrm{NH}), 2.11\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right)$ and $0.96\left(6 \mathrm{H}, \mathrm{m}, \mathrm{Me} e_{2} \mathrm{CH}\right)$; $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 209.06$ (C), 157.05 (C), 136.47 (C), $129.01(\mathrm{CH}), 128.66(\mathrm{CH}), 128.16(\mathrm{CH}), 67.70\left(\mathrm{CH}_{2}\right), 65.86$ $(\mathrm{CH}), 33.85(\mathrm{CH}), 19.89(\mathrm{Me})$ and $18.63(\mathrm{Me}) ; m / z(\mathrm{EI}) 267$ ( $\mathrm{MH}^{+}, 4 \%$ ), 266 ( $\mathrm{M}^{+}, 0.2$ ), 206 (15), 162 (13), 155 (8) and 91 (100).
(R)-Ethyl 2-[1-(benzyloxycarbonylamino)-2-methylpropyl]-thiazole-4-carboxylate 9a. Potassium hydrogen carbonate (107 $\mathrm{mg}, 1.07 \mathrm{mmol}$ ) and ethyl bromopyruvate ( $140 \mu 1,1.12 \mathrm{mmol}$ ) were added sequentially to a stirred solution of the above thioamide $\mathbf{8 a}(71 \mathrm{mg}, 0.27 \mathrm{mmol})$ in 1,2-dimethoxyethane ( 1 ml ) at $-20^{\circ} \mathrm{C}$. The reaction was warmed to $-15^{\circ} \mathrm{C}$, stirred for 16 h and filtered through a pad of Celite. The pad was washed with ether ( 1 ml ) and the combined filtrates were evaporated in vacuo. The residue was dissolved in 1,2-dimethoxyethane (1 ml ), cooled to $-20^{\circ} \mathrm{C}$ and a solution of trifluoroacetic acid ( $120 \mu \mathrm{l}, 0.85 \mathrm{mmol}$ ) and 2,6-lutidine ( $210 \mu \mathrm{l}, 1.80 \mathrm{mmol}$ ) in $1,2-$ dimethoxyethane $(1 \mathrm{ml})$ at $-20^{\circ} \mathrm{C}$ was added. The mixture was stirred for 30 min , evaporated in vacuo and partitioned between chloroform ( 30 ml ) and water ( 30 ml ). The organic layer was washed with water ( 30 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, gradient eluting with ethyl acetate-light petroleum ( $1: 4$ to $1: 2$ ), to afford the title compound 9 a ( $96 \mathrm{mg}, 100 \%$ ) as a pale yellow oil (Found: $\mathrm{M}^{+}$, 362.1301. $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}$ requires $M$, 362.1300); $[a]_{\mathrm{D}}^{20}$ $+26.4^{\circ}\left(c \quad 1.26, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3435,3126,2986$, 2971, 2940, 2908, 2875, 1729, 1504, 1392, 1372, 1260, 1098, 1072, 1051, 1021, 910 and $859 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.99(1 \mathrm{H}$, $\mathrm{s}, \mathrm{SCH}), 7.27(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), 5.57\left(1 \mathrm{H}, \mathrm{d}, J 4.5\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$, OCONH), $5.05\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.88(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{NH}), 4.32$ $\left(2 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{OCH}_{2} \mathrm{Me}\right), 2.34\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 1.32(3 \mathrm{H}, \mathrm{t}$, $J 7.1, \mathrm{OCH}_{2} \mathrm{Me}$ ), $0.89(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{MeCHMe})$ and $0.85(3 \mathrm{H}$, d, J6.8, МeСН Me); $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 172.64$ (C), 161.66 (C), 156.46 (C), 147.77 (C), 136.58 (C), 128.91 (CH), 128.57 $(\mathrm{CH}), 128.44(\mathrm{CH}), 127.26(\mathrm{CH}), 67.54\left(\mathrm{CH}_{2}\right), 61.80\left(\mathrm{CH}_{2}\right)$, $59.04(\mathrm{CH}), 33.81(\mathrm{Me}), 19.77(\mathrm{Me}), 17.87(\mathrm{CH})$ and 14.73 (Me); $m / z(\mathrm{EI}) 363\left(\mathrm{MH}^{+}, 14 \%\right), 362\left(\mathrm{M}^{+}, 10\right), 320$ (4), 275 (19), 227 (9), 139 (7) and 91 (100).
(R)-2-[1-(Benzyloxycarbonylamino)-2-methylpropyl]thiazole-4-carboxylic acid 2a. Sodium hydroxide ( $42 \mathrm{mg}, 1.05 \mathrm{mmol}, 5.2$ equiv.) was added in one portion to a stirred solution of the ester $\mathbf{9 a}(72 \mathrm{mg}, 0.20 \mathrm{mmol})$ in THF-water (3:1) ( 6 ml ) at room
temperature. The mixture was stirred for 5 h , the THF was evaporated in vacuo and the mixture was partitioned between ether ( 25 ml ) and water ( 25 ml ). The aqueous layer was acidified to pH 2 with 1 m aqueous hydrochloric acid and extracted with ethyl acetate ( $2 \times 20 \mathrm{ml}$ ). The combined organic extracts were washed with brine ( 25 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo to afford the title compound 2a (55 $\mathrm{mg}, 83 \%)$ as a colourless oil; $[a]_{\mathrm{D}}^{24}+27.2^{\circ}\left(c 1.18, \mathrm{CHCl}_{3}\right)$; $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3315,2965,2934,1706,1516,1232,1098$, 1026, 753 and $697 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.13(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH})$, $7.46\left(1 \mathrm{H}, \mathrm{br}\right.$ s, $\left.\mathrm{CO}_{2} \mathrm{H}\right), 7.27(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), 6.42(0.17 \mathrm{H}, \mathrm{br}$ s, OCONH), $5.67\left(0.83 \mathrm{H}, \mathrm{d}, J 8.2\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OCONH}\right), 5.07$ $(1 \mathrm{H}, \mathrm{d}, J 12.5, \mathrm{PhC} H \mathrm{H}), 5.04(1 \mathrm{H}, \mathrm{d}, J 12.5, \mathrm{PhCH} H), 4.89$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{NH}$ ), $2.34\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right.$ ), 0.89 ( $3 \mathrm{H}, \mathrm{d}, J 6.8$, $M e \mathrm{CHMe})$ and $0.88(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{MeCH} M e) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) 171.84 (C), 162.90 (C), 155.16 (C), 145.47 (C), 135.10 $(\mathrm{C}), 127.54(\mathrm{CH}), 127.41(\mathrm{CH}), 127.26(\mathrm{CH}), 127.08(\mathrm{CH}), 66.26$ $\left(\mathrm{CH}_{2}\right), 57.62(\mathrm{CH}), 32.34(\mathrm{CH}), 18.40(\mathrm{Me})$ and $16.62(\mathrm{Me})$.

## (R)-2-[1-(tert-Butoxycarbonylamino)-2-methylpropyl]thiazole-4carboxylic acid 26

(R)- $\boldsymbol{N}^{\mathbf{2}}$-(tert-Butoxycarbonyl)valinamide 7b. Prepared in $88 \%$ yield by the method described for the $(S)$-enantiomer in the preceding paper, ${ }^{16} \mathrm{mp} 162-163^{\circ} \mathrm{C}$ (chloroform-light petroleum) [lit., ${ }^{38}(S)$-enantiomer; mp $\left.160-161^{\circ} \mathrm{C}\right] ;[a]_{D}^{24}-16.1^{\circ}$ (c 0.80, DMF) \{lit., ${ }^{16}$ ( $S$ )-enantiomer; $[a]_{D}^{19} 17.7^{\circ}$ (c 1.33, DMF) $\}$.
( $R$ )- $N^{2}$-(tert-Butoxycarbonyl)thiovalinamide $\mathbf{8 b}$. A solution of the above amide $7 \mathbf{b}(0.63 \mathrm{~g}, 2.90 \mathrm{mmol})$ and Lawesson's reagent $(0.61 \mathrm{~g}, 1.50 \mathrm{mmol})$ in dry dichloromethane $(50 \mathrm{ml})$ was stirred for 16 h at room temperature. The mixture was evaporated in vacuo and purified by flash chromatography on silica, eluting with ether-light petroleum ( $1: 2$ ), to afford $(R)-N^{2}$-(tertbutoxycarbonyl)thiovalinamide ( $0.53 \mathrm{~g}, 78 \%$ ) as a pale yellow solid. A small portion was recrystallised to afford the title compound $\mathbf{8 b}$ as colourless prisms, $\mathrm{mp} 101-103^{\circ} \mathrm{C}$ (ethyl acetatelight petroleum) [lit. ${ }^{38}(S)$-enantiomer; $\left.\mathrm{mp} \quad 112-113^{\circ} \mathrm{C}\right]$; $[a]_{D}^{22.5}+44.0^{\circ} \quad\left(c \quad 1.02, \quad \mathrm{CHCl}_{3}\right) \quad\left\{\right.$ lit., ${ }^{38} \quad(S)$-enantiomer; $\left.[a]_{\mathrm{D}}^{25}-43.48^{\circ}\left(c 0.7, \mathrm{CHCl}_{3}\right)\right\} ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3480,3433$, 3367, 3306, 3194, 2976, 2935, 2875, 1700, 1606, 1499, 1394, $1369,1308,1167,1089,1044,1008$ and $871 ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 8.24\left(1 \mathrm{H}, \mathrm{s}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CSNHH}\right), 7.77(1 \mathrm{H}, \mathrm{s}$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CSNH} H\right), 5.35\left(1 \mathrm{H}, \mathrm{d}, J 9\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CHN} H\right), 4.22(1 \mathrm{H}$, $\mathrm{m}, \mathrm{C} H \mathrm{NH}), 2.15\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 1.43\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right)$ and 0.98 ( $6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{Me} e_{2} \mathrm{CH}$ ); $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 209.42(\mathrm{C}), 156.00$ (C), $80.23(\mathrm{C}), 65.09(\mathrm{CH}), 33.32(\mathrm{CH}), 28.32(\mathrm{Me}), 19.47(\mathrm{Me})$ and $18.27(\mathrm{Me}) ; m / z$ (EI) $232\left(\mathrm{M}^{+}, 17 \%\right), 172$ (39), 159 (7), 133 (9), 116 (42), 99 (10), 72 (100), 57 (93) and 41 (41).
( $R$ )-Ethyl 2-[1-(tert-butoxycarbonylamino)-2-methylpropyl]-thiazole-4-carboxylate 9b. Potassium hydrogen carbonate ( 0.75 $\mathrm{g}, 7.45 \mathrm{mmol}$ ) and ethyl bromopyruvate ( $1.0 \mathrm{ml}, 7.97 \mathrm{mmol}$ ) were added sequentially to a stirred solution of the thioamide $\mathbf{8 b}(0.43 \mathrm{~g}, 1.86 \mathrm{mmol}, 1.0$ equiv.) in 1,2-dimethoxyethane ( 4 ml ) at $-40^{\circ} \mathrm{C}$. The reaction was warmed to $-20^{\circ} \mathrm{C}$, stirred for 16 h and filtered through a pad of Celite. The pad was washed with ether ( 4 ml ) and the combined filtrates were evaporated in vacuo. The residue was dissolved in 1,2-dimethoxyethane $(4 \mathrm{ml})$, cooled to $-40^{\circ} \mathrm{C}$ and a solution of trifluoroacetic anhydride ( $0.86 \mathrm{ml}, 6.09 \mathrm{mmol}, 3.3$ equiv.) and 2,6-lutidine ( $1.50 \mathrm{ml}, 12.9 \mathrm{mmol}, 6.9$ equiv.) in 1,2-dimethoxyethane ( 2 ml ) at $-20^{\circ} \mathrm{C}$ was added. The mixture was warmed to $-20^{\circ} \mathrm{C}$ over 10 min , stirred at $-20^{\circ} \mathrm{C}$ for 30 min , evaporated in vacuo and partitioned between chloroform $(40 \mathrm{ml})$ and water $(40 \mathrm{ml})$. The aqueous layer was further extracted with chloroform ( 20 ml ) and the organic extracts were combined, washed with water (40 $\mathrm{ml})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, eluting with ethyl acetate-light petroleum ( $1: 6$ ), to afford the title compound $\mathbf{9 b}(0.49 \mathrm{~g}, 73 \%)$ as colourless needles, mp $118.5-119^{\circ} \mathrm{C}$ (from ethyl acetate-light petroleum) (lit., ${ }^{25} \mathrm{mp} \mathrm{114-115}{ }^{\circ} \mathrm{C}$ ) (Found: C, 54.6; H, 7.5;
$\mathrm{N}, 8.4$. Calc. for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}: \mathrm{C}, 54.9 ; \mathrm{H}, 7.4 ; \mathrm{N}, 8.5 \%$ ); $[a]_{\mathrm{D}}^{19}+41.6^{\circ}\left(c \quad 1.06, \mathrm{CHCl}_{3}\right)\left\{\right.$ lit. ${ }^{25}[a]_{\mathrm{D}}^{25}+39.28^{\circ}$ (c 2.6, $\mathrm{MeOH})\} ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3440,3126,2982,2936,2875$, 1716, 1603, 1498, 1393, 1370, 1165, 1098, 1019, 962 and 870; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.07(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 5.31(1 \mathrm{H}, \mathrm{d}, J 8.0$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CHN} H\right), 4.91(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{NH}), 4.42(2 \mathrm{H}, \mathrm{q}, J 7.2$, $\left.\mathrm{CH}_{2} \mathrm{Me}\right), 2.45\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right), 1.40(3 \mathrm{H}, \mathrm{t}$, $\left.J 7.2, \mathrm{CH}_{2} \mathrm{Me}\right), 0.98(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{MeCHMe})$ and $0.91(3 \mathrm{H}, \mathrm{d}$, $J 6.8, \mathrm{MeCHMe}) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 173.25(\mathrm{C}), 161.38$ (C), 155.45 (C), 147.41 (C), $126.78(\mathrm{CH}), 80.10(\mathrm{C}), 61.41$ $\left(\mathrm{CH}_{2}\right), 58.05(\mathrm{CH}), 36.06(\mathrm{CH}), 28.32(\mathrm{Me}), 19.44(\mathrm{Me}), 17.28$ (Me) and $14.37(\mathrm{Me}) ; m / z(\mathrm{EI}) 329\left(\mathrm{MH}^{+}, 62 \%\right), 328\left(\mathrm{M}^{+}, 1.5\right)$, 285 (17), 273 (70), 229 (47), 185 (74), 139 (40), 72 (11), 57 (100), 41 (42) and 29 (29).
( $R$ )-2-[1-(tert-Butoxycarbonylamino)-2-methylpropyl]thi-azole-4-carboxylic acid 2b. Lithium hydroxide monohydrate (97 $\mathrm{mg}, 2.3 \mathrm{mmol}$ ) was added in one portion to a stirred solution of the ester $9 \mathbf{b}(152 \mathrm{mg}, 0.46 \mathrm{mmol})$ in methanol-water ( $5: 1$ ) $(10 \mathrm{ml})$ at room temperature. The reaction was stirred for 6 h , the methanol was evaporated in vacuo and the mixture was partitioned between ether ( 40 ml ) and water ( 40 ml ). The aqueous layer was acidified to pH 4 with $10 \%$ aqueous citric acid, extracted with ethyl acetate ( $2 \times 40 \mathrm{ml}$ ), acidified to pH 3 with $10 \%$ aqueous citric acid and further extracted with ethyl acetate $(2 \times 40 \mathrm{ml})$. The combined organic extracts were washed with brine ( 75 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo to afford the title compound $\mathbf{2 b}(0.14 \mathrm{~g}$, quant.) as a colourless solid. A small portion was recrystallised to afford colourless prisms, $\mathrm{mp} 300-302^{\circ} \mathrm{C}$ (lit., ${ }^{26} \mathrm{mp}$ not given) (Found: $\mathrm{M}^{+}, 300.1146$. Calc. for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}: ~ M, ~ 300.1144$ ); $[a]_{\mathrm{D}}^{20}+35.0^{\circ}\left(c \quad 1.17, \mathrm{CHCl}_{3}\right)\left\{\right.$ lit., ${ }^{26}(S)$-enantiomer; $[a]_{\mathrm{D}}^{21.5}$ -42.0 (c 2.6, $\left.\left.\mathrm{CHCl}_{3}\right)\right\} ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3440,3124,2979$, 2934, 2875, 1709, 1496, 1393, 1369, 1341, 1240, 1164, 1097, 1042 and $869 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}, 20^{\circ} \mathrm{C}\right) 9.50(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CO}_{2} \mathrm{H}\right), 8.22(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 6.69\left(0.35 \mathrm{H}\right.$, br s, exch. $\mathrm{D}_{2} \mathrm{O}$, CHNH), $5.44\left(0.65 \mathrm{H}, \mathrm{m}, J 8.0\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CHN} H\right), 4.93$ $(0.65 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{NH}), 4.88(0.35 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{NH}), 2.43(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{C} H \mathrm{Me}_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right)$ and $0.97\left(6 \mathrm{H}, \mathrm{m}, M e_{2} \mathrm{CH}\right) ; \delta_{\mathrm{H}}(400$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}, 50^{\circ} \mathrm{C}\right) 8.13(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 5.15(1 \mathrm{H}, \mathrm{m}$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CHNH}\right), 4.85(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{NH}), 2.36\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right)$, $1.44\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right), 0.99(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{MeCHMe})$ and $0.95(3 \mathrm{H}$, d, J6.8, MeCHMe); $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 174.04$ (C), 164.70 $(\mathrm{C}), 155.93(\mathrm{C}), 147.03(\mathrm{C}), 128.68(\mathrm{CH}), 82.42(\mathrm{C}), 58.43(\mathrm{CH})$, $33.66(\mathrm{CH}), 28.68(\mathrm{Me}), 19.79(\mathrm{Me})$ and $17.79(\mathrm{Me}) ; m / z(\mathrm{EI})$ $301\left(\mathrm{MH}^{+}, 2 \%\right), 300\left(\mathrm{M}^{+}, 0.4\right), 245$ (8), 201 (18), 166 (7), 157 (40), 139 (16), 113 (6), 72 (8), 57 (100) and 41 (68).

## 2-[ $N$-(tert-Butoxycarbonyl)aminomethyl]thiazole-4-carboxylic

 acid 3$\boldsymbol{N}^{2}$-(tert-Butoxycarbonyl)glycinamide 10. To a stirred solution of $N$-(tert-butoxycarbonyl)glycine ( $7.01 \mathrm{~g}, 40 \mathrm{mmol}$ ) and triethylamine ( $5.60 \mathrm{ml}, 40 \mathrm{mmol}$ ) in dry THF ( 60 ml ) was added ethyl chloroformate ( $3.90 \mathrm{ml}, 40 \mathrm{mmol}$ ) dropwise at $-10^{\circ} \mathrm{C}$. The mixture was stirred for 25 min and aqueous ammonia $(35 \% ; 10 \mathrm{ml})$ was added. The mixture was stirred at $-10^{\circ} \mathrm{C}$ for 45 min and partitioned between ethyl acetate ( 75 ml ) and water $(60 \mathrm{ml})$. The aqueous layer was extracted with ethyl acetate ( 50 $\mathrm{ml})$ and the organic extracts were combined, washed sequentially with aqueous sodium hydrogen carbonate ( 75 ml ), brine ( $2 \times 75 \mathrm{ml}$ ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, eluting with ethyl acetatelight petroleum ( $4: 1$ ), to afford the title compound $\mathbf{1 0}(1.99 \mathrm{~g}$, $29 \%$ ) as a colourless solid. A small portion was recrystallised to afford the title compound as colourless prisms, $\mathrm{mp} 85-86^{\circ} \mathrm{C}$ (ethyl acetate-light petroleum) (lit., ${ }^{28} \mathrm{mp} 94^{\circ} \mathrm{C}$ ) (Found: C, 48.4; H, 8.3; $\mathrm{N}, 15.8$. Calc. for $\mathrm{C}_{7} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3}$ : C, 48.3; H, 8.1; N , $16.1 \%) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3524,3451,3410,2983,2935,2872$, 1691, $1593,1576,1504,1454,1394,1369$ and 1164; $\delta_{\mathrm{H}}(400$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 6.53\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CONHH}\right), 6.30(1 \mathrm{H}$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CONH} H\right), 5.56\left(1 \mathrm{H}\right.$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CH}_{2} \mathrm{~N} H\right)$,
$3.81\left(2 \mathrm{H}, \mathrm{d}, J 4.3, \mathrm{CH}_{2} \mathrm{NH}\right)$ and $1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right) ; \delta_{\mathrm{C}}(100.6$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 173.13(\mathrm{C}), 156.60(\mathrm{C}), 80.57(\mathrm{C}), 44.18\left(\mathrm{CH}_{2}\right)$ and $28.70(\mathrm{Me}) ; m / z(\mathrm{EI}) 175\left(\mathrm{MH}^{+}, 40 \%\right), 174\left(\mathrm{M}^{+}, 0.1\right), 130$ (14), 119 (90), 101 (32), 75 (38), 57 (100), 41 (57) and 30 (98).
$\boldsymbol{N}^{2}$ (tert-Butoxycarbonyl)thioglycinamide 11. A solution of the above amide $\mathbf{1 0}(0.96 \mathrm{~g}, 5.49 \mathrm{mmol})$ and Lawesson's reagent $(1.16 \mathrm{~g}, 2.86 \mathrm{mmol})$ in dry dichloromethane $(65 \mathrm{ml})$ was stirred for 16 h at room temperature. The mixture was evaporated in vacuo and purified by flash chromatography on silica, gradient eluting with ether-light petroleum ( $1: 1$ to $3: 1$ ), to afford the title compound $11(0.89 \mathrm{~g}, 85 \%)$ as a colourless solid. A small portion was recrystallised to afford colourless prisms, mp 126$127^{\circ} \mathrm{C}$ (ethyl acetate-light petroleum) (lit. ${ }^{28} \mathrm{mp} 124^{\circ} \mathrm{C}$ ); $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3480,3450,3364,2983$, 2935, 2874, 1719, 1706, 1606, 1595, 1504, 1422, 1370, 1251, 1162, 1047, 914 and $856 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.52\left(1 \mathrm{H}, \mathrm{br}\right.$ s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CSNH} H\right)$, $8.30\left(1 \mathrm{H}\right.$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CSNH} H\right), 5.89\left(1 \mathrm{H}\right.$, br s, exch. $\mathrm{D}_{2} \mathrm{O}$, $\left.\mathrm{CH}_{2} \mathrm{~N} H\right), 4.10\left(2 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CH}_{2} \mathrm{NH}\right)$ and $1.46\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right)$; $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ;\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 204.50(\mathrm{C}), 155.98$ (C), 78.77 (C), $50.92\left(\mathrm{CH}_{2}\right)$ and $28.61(\mathrm{Me}) ; m / z(\mathrm{EI}) 190\left(\mathrm{M}^{+}, 35 \%\right), 135(51)$, 134 (50), 117 (19), 75 (47), 57 (100), 41 (62) and 30 (51).
Ethyl 2-[ $\boldsymbol{N}$-(tert-butoxycarbonyl)aminomethyl]thiazole-4carboxylate 12. Potassium hydrogen carbonate ( $0.75 \mathrm{~g}, 7.45$ $\mathrm{mmol})$ and ethyl bromopyruvate ( $1.0 \mathrm{ml}, 7.97 \mathrm{mmol}$ ) were added sequentially to a stirred solution of the thioamide $\mathbf{1 1}$ ( $0.43 \mathrm{~g}, 1.86 \mathrm{mmol}$ ) in 1,2 -dimethoxyethane ( 4 ml ) at $-25^{\circ} \mathrm{C}$. The reaction was warmed to $-15^{\circ} \mathrm{C}$, stirred for 16 h and filtered through a pad of Celite. The pad was washed with diethyl ether $(4 \mathrm{ml})$ and the combined filtrates were evaporated in vacuo. The residue was dissolved in 1,2-dimethoxyethane $(4 \mathrm{ml})$, cooled to $-10^{\circ} \mathrm{C}$ and a solution of trifluoroacetic anhydride ( $0.86 \mathrm{ml}, 6.09 \mathrm{mmol}$ ) and 2,6-lutidine ( $1.50 \mathrm{ml}, 12.9$ mmol ) in 1,2 -dimethoxyethane ( 2 ml ) at $-10^{\circ} \mathrm{C}$ was added. The mixture was stirred at $-10^{\circ} \mathrm{C}$ for 30 min , evaporated in vacuo and partitioned between chloroform ( 35 ml ) and water $(35 \mathrm{ml})$. The aqueous layer was further extracted with chloroform ( 35 ml ) and the organic extracts were combined, washed with water $(45 \mathrm{ml})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, eluting with ethyl acetate-light petroleum (1:2), to afford the title compound $\mathbf{1 2}$ $(0.46 \mathrm{~g}, 61 \%)$ as a pale yellow solid. A small portion was recrystallised to afford colourless needles, $\mathrm{mp} 103-104{ }^{\circ} \mathrm{C}$ (ethyl acetate-light petroleum) (lit., ${ }^{28} \mathrm{mp} 133{ }^{\circ} \mathrm{C}$ ) (Found: C, 50.4; H, 6.5; $\mathrm{N}, 9.9$. Calc. for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}: \mathrm{C}, 50.3 ; \mathrm{H}, 6.3$; N, $9.8 \%$ ); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3455,3125,2984,2938,2873,1718,1603$, 1503, 1456, 1447, 1370, 1320, 1166, 1099, 1022, 930, 919 and $858 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.13(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 5.94(1 \mathrm{H}$, m , exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CH}_{2} \mathrm{NH}\right), 4.65\left(2 \mathrm{H}, \mathrm{d}, J 6.3, \mathrm{CH}_{2} \mathrm{NH}\right), 4.40$ $\left(2 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{CH}_{2} \mathrm{Me}\right), 1.46\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right)$ and $1.39(3 \mathrm{H}$, t , J 7.1, $\left.\mathrm{CH}_{2} \mathrm{Me}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 170.97$ (C), 161.57 (C), 156.16 (C), 147.06 (C), $128.15(\mathrm{CH}), 80.49$ (C), 61.69 $\left(\mathrm{CH}_{2}\right), 42.70\left(\mathrm{CH}_{2}\right), 28.59(\mathrm{Me})$ and $14.80(\mathrm{Me}) ; m / z$ (EI) $287\left(\mathrm{MH}^{+}, 45 \%\right), 286\left(\mathrm{M}^{+}, 3\right), 231$ (100), 213 (31), 187 (28), 167 (15), 158 (18), 140 (20), 112 (37), 57 (100), 41 (48) and 29 (48).

## 2-[ $N$-(tert-Butoxycarbonyl)aminomethyl]thiazole-4-carb-

oxylic acid 3. Lithium hydroxide monohydrate $(85 \mathrm{mg}, 2.0$ $\mathrm{mmol})$ was added in one portion to a stirred solution of the thiazole ester $\mathbf{1 2}(116 \mathrm{mg}, 0.40 \mathrm{mmol})$ in methanol-water ( $5: 1$ ) $(10 \mathrm{ml})$ at room temperature. The reaction was stirred for 3 h , the methanol was evaporated in vacuo and the mixture was partitioned between ether ( 30 ml ) and water ( 30 ml ). The aqueous layer was acidified to pH 3.5 with $10 \%$ aqueous citric acid, extracted with ethyl acetate ( $2 \times 30 \mathrm{ml}$ ), acidified to pH 2 with $10 \%$ aqueous citric acid and further extracted with ethyl acetate $(2 \times 40 \mathrm{ml})$. The combined organic extracts were washed with brine ( 60 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo to afford the title compound $\mathbf{3}(91 \mathrm{mg}, 87 \%)$ as a colourless solid. A small portion was recrystallised to afford colourless prisms, mp $180.5-182{ }^{\circ} \mathrm{C}$ (ethyl acetate-light petroleum) (lit. ${ }^{28} \mathrm{mp} 184^{\circ} \mathrm{C}$ )
(Found: C, 46.5; H, 5.4; N, 10.7. Calc. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}: \mathrm{C}$, $46.5 ; \mathrm{H}, 5.5 ; \mathrm{N}, 10.85 \%) ; v_{\max }\left(\mathrm{CHCl}_{3} / \mathrm{cm}^{-1} 3455,3125,3038\right.$, 2984, 2934, 1715, 1506, 1369 and 1165; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$, $\left.20^{\circ} \mathrm{C}\right) 8.12(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 7.20\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CO}_{2} \mathrm{H}\right)$, $5.99\left(0.2 \mathrm{H}\right.$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CH}_{2} \mathrm{~N} H\right)$, $5.75(0.8 \mathrm{H}$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CH}_{2} \mathrm{NH}\right), 4.64\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{NH}\right)$ and $1.46\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right)$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}, 50^{\circ} \mathrm{C}\right) 8.18(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 5.20(1 \mathrm{H}$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CH}_{2} \mathrm{~N} H\right), 4.61\left(2 \mathrm{H}, \mathrm{d}, J 6.3, \mathrm{CH}_{2} \mathrm{NH}\right)$ and 1.47 $\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 170.37(\mathrm{C}), 163.32(\mathrm{C})$, $156.12(\mathrm{C}), 147.76(\mathrm{C}), 128.08(\mathrm{CH}), 80.57(\mathrm{C}), 42.68\left(\mathrm{CH}_{2}\right)$ and $28.67(\mathrm{Me}) ; m / z(\mathrm{EI}) 259\left(\mathrm{MH}^{+}, 19 \%\right), 258\left(\mathrm{M}^{+}, 2\right)$, 203 (77), 185 (32), 167 (15), 159 (27), 141 (11), 57 (100), 41 (49) and 29 (24).

## Coupling reactions

Methyl 2-[1-(\{2-[1-(benzyloxycarbonylamino)-2-methylpropyl]-thiazol-4-yl\}carbonylamino)ethyl]-5-methyloxazole-4-carboxylate 13a
$N$-Methylmorpholine ( $22 \mu \mathrm{l}, 0.20 \mathrm{mmol}$ ) and isobutyl chloroformate ( $26 \mu \mathrm{l}, 0.20 \mathrm{mmol}$ ) were added sequentially to a stirred solution of the thiazole carboxylic acid $\mathbf{2 a}(52 \mathrm{mg}, 0.159 \mathrm{mmol})$ in dry THF ( 5 ml ) at $0^{\circ} \mathrm{C}$. The mixture was stirred for 30 min at $0^{\circ} \mathrm{C}$, a solution of the oxazole $\mathbf{1}(32 \mathrm{mg}, 0.174 \mathrm{mmol})$ in dry THF ( 4 ml ) was added, the reaction was stirred for 45 min and partitioned between ethyl acetate $(30 \mathrm{ml})$ and water $(30 \mathrm{ml})$. The aqueous layer was further extracted with ethyl acetate ( 20 ml ) and the organic extracts were combined, washed sequentially with saturated aqueous sodium hydrogen carbonate $(50 \mathrm{ml})$ and brine ( 50 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, eluting with ethyl acetatelight petroleum ( $1: 1$ ), to afford the title compound $\mathbf{1 3 a}$ ( 55 mg , $69 \%$ ) as a colourless oil (Found: M ${ }^{+}$, 500.1728. $\mathrm{C}_{24} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{6} \mathrm{~S}$ requires $M, 500.1729) ;[a]_{\mathrm{D}}^{19}+36.0^{\circ}\left(c 1.05, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) /$ $\mathrm{cm}^{-1} 3436,3401,3125,3091,3064,3050,2967,2936,2877$, 1723, 1669, 1540, 1505, 1352 and 1100; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $8.04(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 7.71\left(1 \mathrm{H}, \mathrm{d}, J 8.5\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CCONH}\right), 7.36$ $(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), 5.53\left[1 \mathrm{H}, \mathrm{d}, J 8.4\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OC}(\mathrm{O}) \mathrm{NH}\right], 5.45$ ( $1 \mathrm{H}, \mathrm{dq}, J 8.5,7.1, \mathrm{NHCHMe}), 5.14\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.94(1 \mathrm{H}$, dd, $J 8.4,5.5, \mathrm{CHC} H \mathrm{NH}$ ), 3.89 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), 2.61 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Me}$ ), $2.35(1 \mathrm{H}, \mathrm{m}, \mathrm{CHMe}), 1.67(3 \mathrm{H}, \mathrm{d}, J 7.1, \mathrm{NHCHMe}), 0.98(3 \mathrm{H}$, d, J6.8, MeCHMe) and 0.93 (3H, d, J6.9, MeCHMe); $\delta_{\mathrm{C}}(100.6$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 172.02 (C), 162.59 (C), 162.31 (C), 160.27 (C), 156.67 (C), 156.06 (C), 149.41 (C), 136.15 (C), 128.59 (CH), $128.31(\mathrm{CH}), 128.19(\mathrm{CH}), 127.49(\mathrm{C}), 123.59(\mathrm{CH}), 67.31$ $\left(\mathrm{CH}_{2}\right), 58.59(\mathrm{CH}), 51.95(\mathrm{Me}), 42.93(\mathrm{CH}), 33.46(\mathrm{CH}), 19.70$ (Me), $19.36(\mathrm{Me}), 17.55(\mathrm{Me})$ and $12.02(\mathrm{Me}) ; m / z$ (EI) 500 $\left(\mathrm{M}^{+}, 7 \%\right), 183$ (26), 168 (15), 113 (12), 91 (100) and 51 (28).

Methyl 2-[1-(\{2-[1-(tert-butoxycarbonylamino)-2-methylpropyl]-thiazol-4-yl\}carbonylamino)ethyl]-5-methyloxazole-4-carboxylate 13b
$N$-Methylmorpholine ( $34 \mu \mathrm{l}, 0.31 \mathrm{mmol}$ ) and isobutyl chloroformate ( $40 \mu \mathrm{l}, 0.31 \mathrm{mmol}$ ) were added sequentially to a stirred solution of the thiazole carboxylic acid $\mathbf{2 b}$ ( $75 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) in dry THF ( 4 ml ) at $0^{\circ} \mathrm{C}$. The mixture was stirred for 30 min at $0^{\circ} \mathrm{C}$, a solution of the oxazole $\mathbf{1}(65 \mathrm{mg}, 0.36 \mathrm{mmol})$ in dry THF $(3 \mathrm{ml})$ was added, the reaction was stirred for 1 h and partitioned between ethyl acetate $(30 \mathrm{ml})$ and water $(30 \mathrm{ml})$. The aqueous layer was further extracted with ethyl acetate ( 20 ml ) and the organic extracts were combined, washed sequentially with saturated aqueous sodium hydrogen carbonate ( 50 ml ) and brine ( 50 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, eluting with ethyl acetatelight petroleum ( $1: 2$ ), to afford the title compound $\mathbf{1 3 b}(84 \mathrm{mg}$, $71 \%$ ) as a colourless oil (Found: $\mathrm{MH}^{+}$, 467.1964. $\mathrm{C}_{21} \mathrm{H}_{30^{-}}$ $\mathrm{N}_{4} \mathrm{O}_{6} \mathrm{~S}+\mathrm{H}$ requires 467.1964); $[a]_{\mathrm{D}}^{23}+37.7^{\circ}\left(c 1.10, \mathrm{CHCl}_{3}\right)$; $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3444,3400,3124,2977,2935,2876,1718$, $1670,1623,1589,1540,1494,1443,1392,1369,1353,1165$ and $1100 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.04(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 7.72(1 \mathrm{H}, \mathrm{d}$,
$J 8.5, \mathrm{CCONH}), 5.46(1 \mathrm{H}, \mathrm{dq}, J 8.5,7.1, \mathrm{NHCHMe}), 5.20(1 \mathrm{H}$, d, $J 8.5$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OCON} H\right), 4.87(1 \mathrm{H}, \mathrm{m}, \mathrm{NHCHCHMe})_{2}$, $3.91(3 \mathrm{H}, \mathrm{s}, \mathrm{MeO}), 2.62(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.35\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 1.69$ ( $3 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{NHCHMe}), 1.47\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right), 1.00(3 \mathrm{H}, \mathrm{d}, J 6.7$, $M e \mathrm{CHMe})$ and 0.93 (3H, d, J 6.9, MeCHMe); $\delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 173.03(\mathrm{C}), 162.61(\mathrm{C}), 162.32(\mathrm{C}), 160.35(\mathrm{C}), 156.71$ (C), 155.45 (C), 149.35 (C), 127.45 (C), 123.45 (CH), 80.27 (C), $58.05(\mathrm{CH}), 51.99(\mathrm{Me}), 42.83(\mathrm{CH}), 33.28(\mathrm{CH}), 28.34(\mathrm{Me})$, $19.71(\mathrm{Me}), 19.40(\mathrm{Me}), 17.38(\mathrm{Me})$ and $12.04(\mathrm{Me}) ; \mathrm{m} / z(\mathrm{CI})$ 323 ( $8 \%$ ), 209 (5), 183 (13), 168 (14), 139 (17), 112 (17), 57 (47) and 41 (100).

Methyl 2-(1-\{[2-(1-amino-2-methylpropyl)thiazol-4-yl]carbonyl-amino\}ethyl)-5-methyloxazole-4-carboxylate hydrochloride 14
Acetyl chloride ( $150 \mu \mathrm{l}, 2.11 \mathrm{mmol}$ ) was added portionwise over 2.5 h to a stirred solution of the above amide $\mathbf{1 3 b}(84 \mathrm{mg}, 0.18$ $\mathrm{mmol})$ in dry methanol $(2 \mathrm{ml})$ at room temperature. The reaction was stirred for 2 h and evaporated in vacuo to afford the amine hydrochloride $\mathbf{1 4}(72 \mathrm{mg}, 100 \%)$ as a colourless foam (Found: $[\mathrm{M}-\mathrm{HCl}]^{+}$, 366.1361. $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{4} \mathrm{~S}$ requires [ $M-$ $\left.\mathrm{HCl}]^{+}, 366.1362\right) ;[a]_{\mathrm{D}}^{23}+13.1^{\circ}(c 1.29, \mathrm{MeOH}) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) /$ $\mathrm{cm}^{-1} 3243,3123,2970,2879,2674,1726,1669,1623,1544$, 1496, 1444, 1389, 1374, 1353, 1100, 1036, 1011 and 983; $\delta_{\mathrm{H}}(400$ $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) 8.82\left(1 \mathrm{H}, \mathrm{s}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CONH}\right), 8.39(1 \mathrm{H}, \mathrm{s}$, $\mathrm{SCH}), 5.36(1 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{NHCHMe}), 4.72(1 \mathrm{H}, \mathrm{d}, J 6.4$, $\mathrm{NH}_{2} \mathrm{CHCHMe}$ ), $3.89(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.62(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.47(1 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 1.69(3 \mathrm{H}, \mathrm{d}, J 7.1, \mathrm{NHCHMe}), 1.16(3 \mathrm{H}, \mathrm{d}, J 6.9$, $M e \mathrm{CHMe})$ and 1.05 (3H, d, J 6.7, MeCHMe); $\delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\mathrm{CD}_{3} \mathrm{OD}\right) 163.03$ (C), 160.98 (C), 160.23 (C), 158.89 (C), 154.87 (C), 146.59 (C), 124.68 (C), $124.18(\mathrm{CH}), 55.76$ (CH), 49.43 (Me), 46.69 (Me), $41.47(\mathrm{CH}), 30.45(\mathrm{CH}), 15.83(\mathrm{Me}), 15.48$ (Me) and $9.05(\mathrm{Me}) ; m / z(\mathrm{EI}) 366$ ([M - HCl] $\left.]^{+}, 2.5 \%\right), 323$ (54), 182 (23), 168 (100), 139 (47), 112 (24), 72 (87), 55 (42), 43 (58) and 31 (38).

2-[1-(\{2-[1-(\{2-[(tert-Butoxycarbonylamino)methyl]thiazol-4-yl\}-carbonylamino)-2-methylpropyl]thiazol-4-yl\}carbonylamino)-ethyl]-5-methyloxazole-4-carboxylic acid 16
$N$-Methylmorpholine ( $35 \mu 1,0.32 \mathrm{mmol}$ ) and isobutyl chloroformate ( $21 \mu \mathrm{l}, 0.16 \mathrm{mmol}$ ) were added sequentially to a stirred solution of the thiazole carboxylic acid $\mathbf{3}(36 \mathrm{mg}, 0.14 \mathrm{mmol})$ in dry THF ( 2.5 ml ) at $0^{\circ} \mathrm{C}$. The mixture was stirred for 30 min at $0^{\circ} \mathrm{C}$, a solution of the amine hydrochloride $\mathbf{1 4}(56 \mathrm{mg}, 0.14$ $\mathrm{mmol})$ in dry THF ( 2 ml ) was added, the reaction was stirred for 45 min and partitioned between ethyl acetate $(20 \mathrm{ml})$ and water ( 15 ml ). The aqueous layer was further extracted with ethyl acetate ( 20 ml ) and the organic extracts were combined, washed sequentially with saturated aqueous sodium hydrogen carbonate ( 25 ml ) and brine ( 25 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, eluting with ethyl acetate-light petroleum ( $2: 1$ ) to afford the ester $15(60 \mathrm{mg}, 72 \%)$ as a colourless solid, $\mathrm{mp} 84.5-86^{\circ} \mathrm{C}$; $[a]_{\mathrm{D}}^{21.5}+23.8^{\circ}\left(c 1.15, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3456,3399$, 3124, 2968, 2935, 2875, 1720, 1670, 1622, 1603, 1538, 1493, $1443,1392,1369,1353,1166,1101$ and $909 ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 8.00(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 7.98(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 7.81(1 \mathrm{H}, \mathrm{d}, J 9.2$, CONHCHCH), 7.71 ( $1 \mathrm{H}, \mathrm{d}, J 8.5$, CONHCHMe), $5.51(1 \mathrm{H}, \mathrm{br}$ s , exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OCONH}\right), 5.38(1 \mathrm{H}, \mathrm{dq}, J 8.5,7.2$, NHCHMe), $5.25(1 \mathrm{H}, \mathrm{dd}, J 9.2,6.2$, NHCHCHMe 2 ), $4.56(2 \mathrm{H}, \mathrm{d}, J 4.9$, $\left.\mathrm{NHCH}_{2}\right), 3.83(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.55(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.44(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C} H \mathrm{Me}_{2}$ ), $1.62(3 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{NHCHMe}), 1.39\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right)$, $0.96(3 \mathrm{H}, \mathrm{d}, J 6.7, \mathrm{Me} \mathrm{CHMe})$ and $0.95(3 \mathrm{H}, \mathrm{d}, J 6.8$, $\mathrm{MeCH} M e) ; \delta_{\mathrm{c}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 172.79$ (C), 171.46 (C), 163.88 (C), 163.64 (C), 162.08 (C), 161.57 (C), 157.96 (C), 157.00 (C), 150.61 (C), 150.32 (C), 128.70 (C), 125.68 (CH), $124.95(\mathrm{CH}), 81.79(\mathrm{C}), 57.64(\mathrm{CH}), 53.27(\mathrm{Me}), 44.10(\mathrm{CH})$, $43.66\left(\mathrm{CH}_{2}\right), 34.55(\mathrm{CH}), 29.59(\mathrm{Me}), 20.96(\mathrm{Me}), 20.89(\mathrm{Me})$, $19.13(\mathrm{Me})$ and $13.34(\mathrm{Me}) ; m / z(\mathrm{EI}) 606\left(\mathrm{M}^{+}, 0.1 \%\right)$, 506 (3), 463 (3), 434 (3), 199 (4), 183 (10), 168 (10), 139 (15), 112 (18), 84 (11), 59 (84) and 41 (100).

A solution of lithium hydroxide monohydrate $(40 \mathrm{mg}, 0.95$ mmol ) in water ( 1.5 ml ) was added to a solution of peptide $\mathbf{1 5}$ ( $54 \mathrm{mg}, 88.5 \mu \mathrm{~mol}$ ) in methanol ( 5 ml ) portionwise over 30 h at room temperature. The reaction was stirred overnight, evaporated in vacuo and partitioned between water ( 30 ml ) and ether ( 30 ml ). The aqueous layer was separated, acidified to pH 5 with $10 \%$ aqueous citric acid, extracted with ethyl acetate $(2 \times 30 \mathrm{ml})$, acidified to pH 3 with $10 \%$ aqueous citric acid and further extracted with ethyl acetate $(2 \times 30 \mathrm{ml})$. The organic layers were combined, washed with brine ( 60 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo to afford the title compound $16(49 \mathrm{mg}, 93 \%)$ as a colourless solid, $\mathrm{mp} 118.5-120^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 592.1807. $\mathrm{C}_{25} \mathrm{H}_{32} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{~S}_{2}$ requires $M$, 592.1774); $[a]_{\mathrm{D}}^{24}$ $+20.3^{\circ}\left(c 1.01, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3455,3399,3124$, 2966, 2934, 1715, 1670, 1623, 1603, 1538, 1494, 1454, 1394, 1369, 1167, 1102 and $909 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}, 20^{\circ} \mathrm{C}\right) 8.12$ $(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 8.10(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 7.94\left(2 \mathrm{H}, \mathrm{m}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$, $2 \times \mathrm{CCONH}), 6.83\left(1 \mathrm{H}\right.$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CO}_{2} \mathrm{H}\right), 6.63(0.3 \mathrm{H}$, br s , exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OCONH}\right), 5.61\left(0.7 \mathrm{H}\right.$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OCONH}\right)$, $5.47(1 \mathrm{H}, \mathrm{m}, \mathrm{NHC} H \mathrm{Me}), 5.33(1 \mathrm{H}, \mathrm{dd}, J 9.0,6.3$, NHCH$\mathrm{CHMe}_{2}$ ), $4.63\left(2 \mathrm{H}, \mathrm{d}, J 5.2, \mathrm{NHCH}_{2}\right), 2.63(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.49$ $\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 1.69(3 \mathrm{H}, \mathrm{d}, J 7.0$, NHCHMe), $1.46(9 \mathrm{H}, \mathrm{s}$, $\mathrm{CMe}_{3}$ ), $1.03(3 \mathrm{H}, \mathrm{d}, J 6.3, \mathrm{MeCHMe})$ and $1.01(3 \mathrm{H}, \mathrm{d}, J 6.3$, $\mathrm{MeCHMe}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}, 50^{\circ} \mathrm{C}\right) 8.06(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH})$, $8.04(1 \mathrm{H}, \mathrm{s}, \mathrm{SCH}), 7.83\left(1 \mathrm{H}, \mathrm{m}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{CCONH}\right), 7.76(1 \mathrm{H}$, $\mathrm{m}, \mathrm{CCONH}), 5.40(1 \mathrm{H}, \mathrm{m}, \mathrm{NHCHMe}), 5.35(1 \mathrm{H}, \mathrm{dd}, J 9.2,6.2$, NHCHCHMe 2 ), $5.29\left(1 \mathrm{H}\right.$, br s, exch. $\left.\mathrm{D}_{2} \mathrm{O}, \mathrm{OCONH}\right), 4.61$ $\left(2 \mathrm{H}, \mathrm{d}, J 6.0, \mathrm{NHCH}_{2}\right), 2.62(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.47\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right)$, 1.67 ( $3 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{NHCHMe}$ ), $1.49\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right), 1.05(3 \mathrm{H}, \mathrm{d}$, $J 6.3, \mathrm{MeCHMe})$ and $1.04\left(3 \mathrm{H}, \mathrm{d}, J 6.3\right.$, MeCHMe); $\delta_{\mathrm{C}}(100.6$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 171.84 (C), 170.64 (C), 165.09 (C), 162.82 (C), 161.32 (C), 160.89 (C), 157.64 (C), 156.20 (C), 149.59 (C), 149.31 (C), 127.73 (C), 125.01 (CH), 124.35 (CH), 80.99 (C), $56.83(\mathrm{CH}), 43.35(\mathrm{CH}), 42.76\left(\mathrm{CH}_{2}\right), 33.76(\mathrm{CH}), 28.70(\mathrm{Me})$, $19.98(\mathrm{Me}), 19.94(\mathrm{Me}), 18.26(\mathrm{Me})$ and $12.53(\mathrm{Me}) ; m / z(\mathrm{EI})$ $592\left(\mathrm{M}^{+}, 0.2 \%\right), 549$ (2), 519 (3), 492 (11), 449 (13), 420 (9), 351 (4), 232 (9), 185 (40), 154 (32), 141 (37), 112 (48), 86 (57) and 84 (100).

## (+)-Nostocyclamide

1-(3-Dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride ( $16 \mathrm{mg}, 82 \mu \mathrm{~mol}$ ) was added to a stirred solution of the carboxylic acid $16(43 \mathrm{mg}, 72 \mu \mathrm{~mol})$ and pentafluorophenol $(16 \mathrm{mg}$, $87 \mu \mathrm{~mol})$ in dry dichloromethane $(4 \mathrm{ml})$ at $-10^{\circ} \mathrm{C}$. The mixture was warmed slowly to room temperature over 16 h , evaporated in vacuo and partitioned between ethyl acetate $(20 \mathrm{ml})$ and brine $(20 \mathrm{ml})$. The aqueous layer was further extracted with ethyl acetate ( 10 ml ) and the organic extracts were combined, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo to afford crude pentafluorophenyl ester 17 ( 55 mg ) which was used without further purification. A solution of hydrogen chloride in dioxane ( $4.0 \mathrm{~m} ; 3 \mathrm{ml}$ ) was added to a stirred solution of the pentafluorophenyl ester $17(29 \mathrm{mg}, 39 \mu \mathrm{~mol})$ in dry dioxane $(1.5 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min and warmed rapidly to room temperature. After stirring for 3.5 h , the mixture was evaporated in vacuo and dissolved in chloroform ( 100 ml ). Aqueous potassium hydrogen carbonate ( $1 \mathrm{~m} ; 100 \mathrm{ml}$ ) was added and the mixture was shaken vigorously for 5 min and then separated. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated in vacuo and purified by flash chromatography on silica, eluting with ethyl acetate-light petroleum (2:1), to afford (+)nostocyclamide ( $14 \mathrm{mg}, 74 \%$ ) as a colourless solid, $\mathrm{mp} 251-$ $252{ }^{\circ} \mathrm{C}$. A small portion was recrystallised to afford colourless crystals, mp $259-260^{\circ} \mathrm{C}$ (ethyl acetate-light petroleum) (lit., ${ }^{12}$ mp 255.8-256.9 ${ }^{\circ} \mathrm{C}$ ) (Found: $\mathrm{M}^{+}$, 474.1142. $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{~N}_{6} \mathrm{O}_{4} \mathrm{~S}_{2}$ requires $M$, 474.1144); $[a]_{D}^{19}+51.3^{\circ}\left(c 0.84, \mathrm{CHCl}_{3}\right)\left\{\right.$ lit. $^{35}{ }^{35}$ $\left.[a]_{\mathrm{D}}^{20}+25^{\circ}\left(\mathrm{CHCl}_{3}\right)\right\} ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3397,3125,3006,2966$, 2932, 1667, 1642, 1543, 1521, 1497 and 1449; $\delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 8.56(1 \mathrm{H}, \mathrm{d}, J 5.0, \mathrm{~N} H C H M e), 8.48(1 \mathrm{H}, \mathrm{d}, J 9.0$, $\mathrm{N} H \mathrm{CHCH}), 8.43\left(1 \mathrm{H}, \mathrm{m}, \mathrm{N} H \mathrm{CH}_{2}\right), 8.11\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CSCH}\right)$,
$8.08(1 \mathrm{H}, \mathrm{s}, \mathrm{CHCSC} H), 5.57[1 \mathrm{H}, \mathrm{dd}, J 9.0,3.7$, $\mathrm{NHCH}-$ $\left.\mathrm{CH}(\mathrm{Me})_{2}\right], 5.07(1 \mathrm{H}, \mathrm{dq}, J 5.0,6.5$, NHCHMe), $4.92(1 \mathrm{H}, \mathrm{dd}$, $J 17.5,4.8, \mathrm{NHCHH}), 4.71(1 \mathrm{H}, \mathrm{dd}, J 17.5,2.9$, NHCHH), $2.63(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.25\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 1.63(3 \mathrm{H}, \mathrm{d}, J 6.5$, NHCHMe), $0.89(3 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{MeCHMe})$ and $0.87(3 \mathrm{H}, \mathrm{d}$, $J 7.0, \mathrm{MeCH} M e) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right) 8.34(1 \mathrm{H}, \mathrm{s}, \mathrm{CSCH})$, $8.32(1 \mathrm{H}, \mathrm{s}, \mathrm{CSCH}), 5.70(1 \mathrm{H}, \mathrm{d}, J 3.6$, NHCHCHMe 2 ), 5.18 $(1 \mathrm{H}, \mathrm{q}, J 6.7, \mathrm{NHC} H \mathrm{Me}), 5.05(1 \mathrm{H}, \mathrm{d}, J 17.8, \mathrm{NHC} H \mathrm{H}), 4.81$ $(1 \mathrm{H}, \mathrm{d}, J 17.8, \mathrm{NHCH} H), 2.68(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.36(1 \mathrm{H}, \mathrm{m}$, CHMe $)^{\text {) , }} 1.69$ ( $3 \mathrm{H}, \mathrm{d}, J 6.7$, NHCHMe), 0.98 ( $3 \mathrm{H}, \mathrm{d}, J 6.9$, $\mathrm{MeCHMe})$ and $0.94(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{MeCHMe}) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) 169.09 (C), 165.63 (C), $162.00(\mathrm{C}), 161.33$ (C), 160.76 (C), 160.21 (C), 154.16 (C), 149.46 (C), 149.33 (C), 128.73 (C), 124.77 (CH), $124.10(\mathrm{CH}), 56.38(\mathrm{CH}), 45.54(\mathrm{CH}), 41.39$ $\left(\mathrm{CH}_{2}\right), 36.82(\mathrm{CH}), 20.58(\mathrm{Me}), 19.00(\mathrm{Me}), 17.77(\mathrm{Me})$ and $11.98(\mathrm{Me}) ; m / z(\mathrm{EI}) 474\left(\mathrm{M}^{+}, 3 \%\right), 431\left(\left[\mathrm{M}-\mathrm{C}_{3} \mathrm{H}_{7}\right]^{+}, 13\right), 276$ (26), 155 (40), 135 (44), 78 (71), 51 (55), 44 (72) and 31 (100).

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